metal-organic papers

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Yu-Liang Zhang

School of Chemistry and Pharmaceutics, East China University of Science and Technology, Shanghai 200237, People's Republic of China

Correspondence e-mail: yuliang_zhang@sohu.com

Key indicators

Single-crystal X-ray study T = 273 K Mean σ (C–C) = 0.007 Å R factor = 0.046 wR factor = 0.122 Data-to-parameter ratio = 14.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[*N*,*N*'-Bis(2-oxido-1-naphthylmethylidene)propane-1,3-diamine]manganese(II)

In the title mononuclear manganese(II) complex, $[Mn(C_{25}H_{20}N_2O_2)]$, the Mn atom and the central methylene group of the propanediamine fragment of the ligand lie on a mirror plane. The four-coordinate Mn atom binds to two N and two O atoms of the ligand, forming a square-planar geometry.

Comment

Manganese(II) compounds are very important in bioinorganic chemistry (Ciringh *et al.*, 1997; Chen *et al.*, 2003). The structure of a mononuclear manganese(II) complex, (I), is described here.



Atom Mn1 lies on a mirror plane, as does the methylene C13 and its substituent H atoms. Mn1 is four-coordinate, binding to two N and two O atoms of the ligand in a slightly distorted square-planar geometry (Fig. 1). The Mn-N and Mn-O bond lengths (Table 1) are comparable with the corresponding values observed in other manganese(II) complexes (Gallo, Solari, Re *et al.*, 1997; Gallo, Solari, Floriani *et al.*, 1997). The bond angles subtended by *cis* substituents on Mn1 range from 82.17 (19) to 94.3 (2)° (Table 1).

Experimental

Compound (I) was obtained by stirring propane-1,3-diamine (1.0 mmol, 79.2 mg), 2-hydroxy-1-naphthaldehyde (2.0 mmol, 343.5 mg) and manganese(II) acetate tetrahydrate (1.0 mmol, 245.1 mg) in EtOH solution (80 ml). The residue was recrystallized from an EtOH solution, giving brown block-like crystals.

Crystal data

 $[Mn(C_{25}H_{20}N_2O_2)]$ $M_r = 435.37$ Orthorhombic, $Cmc2_1$ a = 30.650 (1) Å b = 8.464 (3) Å c = 7.769 (1) Å V = 2015.5 (8) Å³ Z = 4 $D_x = 1.435$ Mg m⁻³ Mo $K\alpha$ radiation Cell parameters from 1718 reflections $\theta = 2.5-22.6^{\circ}$ $\mu = 0.68 \text{ mm}^{-1}$ T = 273 (2) K Block, brown $0.32 \times 0.28 \times 0.22 \text{ mm}$

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Data collection

Bruker SMART 1000 CCD area-
detector diffractometer
ω scans
Absorption correction: multi-scan
(SADABS; Sheldrick, 1996)
$T_{\min} = 0.812, \ T_{\max} = 0.865$
5125 measured reflections

Refinement

Refinement on F^2
$R[F^2 > 2\sigma(F^2)] = 0.046$
$wR(F^2) = 0.122$
S = 0.97
1977 reflections
139 parameters
H-atom parameters constrained

1977 independent reflections 1541 reflections with $I > 2\sigma(I)$ $R_{int} = 0.044$ $\theta_{max} = 26.5^{\circ}$ $h = -29 \rightarrow 38$ $k = -10 \rightarrow 10$ $l = -9 \rightarrow 9$

 $w = 1/[\sigma^{2}(F_{o}^{2}) + (0.0679P)^{2}]$ where $P = (F_{o}^{2} + 2F_{c}^{2})/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.27 \text{ e } \text{Å}^{-3}$ $\Delta\rho_{min} = -0.35 \text{ e } \text{Å}^{-3}$ Absolute structure: Flack (1983), with 849 Friedel pairs Flack parameter: -0.08 (4)

Table 1

Selected geometric parameters (Å, °).

Mn1-O1	1.842 (3)	Mn1-N1	1.867 (4)
$\begin{array}{c} O1^{i}-Mn1-O1\\ O1^{i}-Mn1-N1 \end{array}$	82.17 (19) 173.75 (14)	$\substack{\text{O1}-\text{Mn1}-\text{N1}\\\text{N1}-\text{Mn1}-\text{N1}^{\text{i}}}$	91.75 (15) 94.3 (2)

Symmetry code: (i) -x, y, z.

All H atoms were constrained to their ideal geometries, with C–H = 0.93-0.97Å and with U_{iso} (H) = $1.2U_{eq}$ (C).

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1998); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997*a*); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997*a*); molecular graphics: *SHELXTL* (Sheldrick, 1997*b*); software used to prepare material for publication: *SHELXTL*.





The structure of (I), showing the atom-numbering scheme, with displacement ellipsoids drawn at the 30% probability level. Unlabelled atoms are related to labelled atoms by the symmetry operator (-x, y, z).

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